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worksheet & Unit 7 Worksheets Answer $3 \text{ SO}_2(\text{g}) + 2 \text{ HNO}_3(\text{aq}) + 2 \text{ H}_2\text{O}(\text{l}) + 3 \text{ H}_2\text{SO}_4(\text{aq}) + 2 \text{ NO}(\text{g})$. 3/3480) ... What is the sign of the enthalpy change for an endothermic reaction? exothermic? ... 16. $2 \text{ H}_2(\text{g}) + \text{O}_2(\text{g}) \rightarrow 2 \text{ H}_2\text{O}(\text{g})$. . The role of a catalyst in a chemical reaction is to.. Enthalpy can be calculated from several sources including: • Stoichiometry. Calorimetry. From tables of standard values. Hess's Law. • Bond energies. Reactants.

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